

of PdCl_2 (40 mg mL^{-1} , 0.665 mmol) at 20–30 °C. The reaction mixture was stirred for 30 min. The precipitate that formed was filtered off, washed with water, and dried in air. Compound **2** was obtained in a yield of 0.3 g (75%). Recrystallization from Me_2CO gave coffee-colored crystals, m.p. 167–168 °C. Found (%): C, 31.67; H, 2.45; Cl, 11.64; N, 9.09. $\text{C}_{16}\text{H}_{16}\text{Cl}_2\text{N}_4\text{O}_{10}\text{Pd}$. Calculated (%): C, 31.94; H, 2.68; Cl, 11.78; N, 9.31. IR, ν/cm^{-1} : 690, 745 (CH in Py); 860 (O–NO₂); 1040, 1145 (C–O); 1270, 1620 (ONO₂); 1365, 1435, 2296, 2931, 2944, (CH₂); 1530, 1580, 1605 (C–C and C–N in Py); 1730 (C=O); 3075 (CH).

X-ray diffraction analysis of compounds 1 and 2. Crystals of **1** are monoclinic, $M = 212.16$, $a = 5.997(4)$ Å, $b = 7.782(2)$ Å, $c = 20.771(10)$ Å, $\beta = 90.66(5)^\circ$, $V = 969.3(5)$ Å³, $d_{\text{calc}} = 1.453(5)$ g cm^{-3} , $\lambda = 1.5418$ Å, space group $P2_1/n$, $Z = 4$.

Crystals of complex **2**, $[\text{PdCl}_2(\text{N}_2\text{O}_5\text{C}_3\text{H}_5)_2]$, are monoclinic, $M = 601.65$, $a = 23.112(25)$ Å, $b = 12.003(3)$ Å, $c = 4.341(6)$ Å, $\beta = 89.72^\circ$, $V = 1204.2(9)$ Å³, $d = 1.659(3)$ g cm^{-3} , $\lambda = 0.70926$ Å, space group $P2_1/a$, $Z = 2$.

Intensities of 464 (for **1**) and 2131 (for **2**) observed unique reflections were measured on a four-circle KM-4 diffractometer (KUMA-Diffraction, Poland) in the range $0.02 < \sin\theta/\lambda < 0.50$ using the $\omega/2\theta$ scanning technique. The structures were solved by direct methods using the SHELX-86 program package on a PC computer. The atomic coordinates in the structure of **1** (see Table 1) were refined by full-matrix least squares to $R = 0.11$. Temperature factors of the nonhydrogen atoms were refined anisotropically. The high value of the R factor is attributable to the fact that the X-ray diffraction data were collected from layered crystals prepared by low-temperature crystallization. We failed to improve the quality of the crystals by using other solvents and temperature conditions of crystallization. The atomic coordinates in the structure of **2** (see Table 2) were

refined by full-matrix least squares using the SHELX-93 program to $R = 0.10$. The nonhydrogen atoms were refined anisotropically, and the H atoms were refined isotropically. When absorption correction was applied using the DIFABS program, the refinement converged to the R factor of 0.079.

This work was financially supported by the International Scientific and Technology Center (Project No. 123-94).

References

1. K. Sakai, *Am. J. Cardiology*, 1989, **63**, 2j–10j.
2. S. S. Liberman and L. N. Yakhontov, *Khim.-farm. Zh.*, 1988, 1046 [*Pharm. Chem. J.*, 1988 (Engl. Transl.)].
3. S. J. Lippard, *Appl. Chem.*, 1987, **59**, 731.
4. A. Sigel and H. Sigel, *Metal Ions in Biological Systems*, Marcel Dekker, New York–Basel–Hong Kong, 1996.
5. US Pat. 4584316, 1986.
6. V. D. Sen', V. A. Golubev, L. M. Volkova, and N. P. Kononova, *J. Inorg. Biochem.*, 1996, **64**, 69.
7. M. Feelish and E. A. Noack, *Eur. J. Pharmacol.*, 1987, **139**, 19.
8. F. V. De Feudus, *Drugs Today*, 1989, **25**, 115.
9. A. R. Galla, *Angew. Chem., Int. Ed. Engl.*, 1993, **32**, 378.
10. J. F. Kerwin, J. R. Lanauster, and P. L. Feldman, *J. Med. Chem.*, 1995, **38**, 4343; *Science*, 1992, **258**, 1862.
11. A. R. Butler and D. L. H. Williams, *Chem. Soc. Rev.*, 1993, **22**, 233.
12. B.-Y. Nawata, N. Terao, T. Terazono, K. Igusa, Y. Yutani, and K. Ochi, *Acta Crystallogr., C*, 1987, **43**, 2460.

Received June 6, 1997;
in revised form January 12, 1998

Glycosylation of betulin acetates with glycals

O. B. Flekhter,* L. A. Baltina, L. V. Spirikhin, I. P. Baikova, and G. A. Tolstikov

Institute of Organic Chemistry, Ufa Research Center of the Russian Academy of Sciences,
71 prosp. Oktyabrya, 450054 Ufa, Russian Federation.
Fax: +7 (347 2) 35 6066

Betulin 2-deoxy- α -D-, 2-deoxy- α -L-, and 2,6-dideoxy- α -L-arabino-hexopyranosides were synthesized by acid-catalyzed glycosylation (cationite in the H^+ form, LiBr) of betulin 3- and 28-monoacetates with glycal acetates.

Key words: 3-O-acetylbetulin, 28-O-acetylbetulin, glycal acetates, stereoselective glycosylation, acid catalysis, betulin 2-deoxy- α -D-, 2-deoxy- α -L-, and 2,6-dideoxy- α -L-arabino-hexopyranosides.

The birch bark is rich in pentacyclic triterpenoids. The content of betulin (**1**) in it reaches 35–40%, depending on the species.¹ Derivatives of betulin and betulinic acid have a broad range of biological activity, including antiviral² and antitumor³ action.

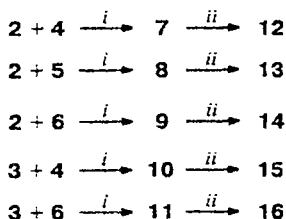
We carried out the glycosylation of betulin monoacetates (**2**, **3**) with acetylated glycals (**4**–**6**) under acid catalysis conditions, using the KU-2-8 cationite (H^+ form) in combination with LiBr as the activator (Scheme 1). We have used this method of glycosylation⁴

Translated from *Izvestiya Akademii Nauk. Seriya Khimicheskaya*, No. 3, pp. 531–534, March, 1998.

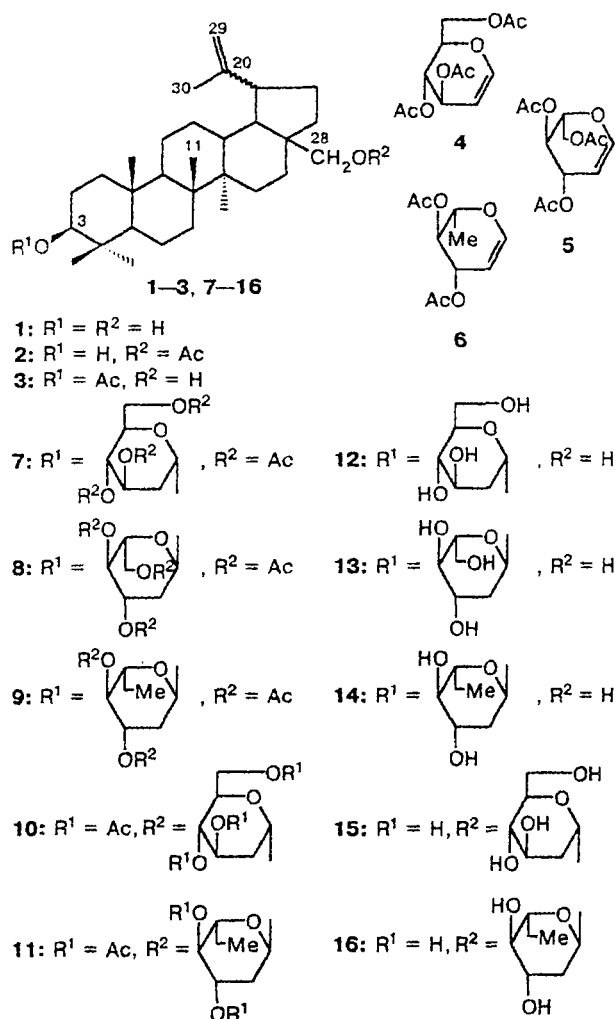
1066-5285/98/4703-0513 \$20.00 © 1998 Plenum Publishing Corporation

for the synthesis of 2-deoxy- α -glycosides of triterpene alcohols of the oleanane series and some steroids.^{5,6} It is necessary to note that the glycosylation of betulin monoacetates with acetobromoglucose gave a mixture of α - and β -glycosides and was accompanied by rearrangements in the aglycone.^{7,8}

Scheme 1



Reagents: i. KU-2-8 cationite (H^+), LiBr;
ii. 5% KOH/MeOH.



The glycosylation of 28-*O*-acetylbetulin 2 with the acetates of D-glucal (4), L-glucal (5), and L-ramnal (6)

gave acetylated 2-deoxy- α -D-, 2-deoxy- α -L, and 2,6-dideoxy- α -L-*arabino*-hexopyranosides (7-9) in 78-82% yields. The glycosylation of the primary alcohol group of 3-*O*-acetylbetulin (3) with the acetates of D-glucal (4) and L-ramnal (6) gave 2-deoxy- α -D- (10) and 2,6-dideoxy- α -L-*arabino*-hexopyranosides (11) in a higher yield (~90%). Treatment of compounds 7-11 with methanolic KOH resulted in the formation of betulin 2-deoxy- α -glycosides (12-16) incorporating a carbohydrate substituent at position 3 or 28.

The structures of the compounds synthesized (7-16) were established by 1H NMR (see Experimental) and ^{13}C NMR spectroscopy (Table 1). Literature data for betulin, its acetates,^{8,9} and 2-deoxy- α -glycosides^{10,11} were used for comparison. The introduction of a carbohydrate residue in the molecule of 28-*O*-acetylbetulin results in a downfield shift of the C(3) signal by 4-10 ppm, while the signals of anomeric C(1') carbon atoms are observed at δ 93.1 and 99.1-99.5, as in the spectra of 2-deoxy- α -glycosides of glycyrrhetic acid and allobetulin.^{10,11} The formation of an α -glycoside bond follows from the magnitudes of the coupling constants, $^1J_{C(1'),H(1')} = 168.3-170.0$ Hz, in the ^{13}C NMR spectra of compounds 7-9.¹² The exact assignment of the signals of carbon and hydrogen atoms in the ^{13}C NMR and 1H NMR spectra of glycoside 8 was carried out by two-dimensional heteronuclear ^{13}C - 1H COSY NMR spectrum. It follows from this spectrum that the signal of the H(1') proton at δ 4.93 correlates with the signal of the C(1') carbon at δ 99.5, while the signals of olefin protons H(29) at δ 4.51 and 4.63 have a cross-peak with the C(29) signal at δ 109.8. The signal of the C(3) aglycone atom at δ 89.3 correlates with the signal of the H(3) atom at δ 2.98. The mutual arrangement of hydrogen atoms determined from 1H NMR and ^{13}C - 1H COSY spectra indicates that glycoside 8 has the structure of 2-deoxy- α -L-*arabino*-hexopyranoside.

The formation of the glycoside bond in 3-*O*-acetylbetulin through the primary hydroxyl group resulted in a downfield shift of the C(28) signal by 6 and 10 ppm, respectively, in the spectra of compounds 10 and 11. The signal of the anomeric C(1') atom in the spectrum of glycoside 10 is observed at δ 97.7, as in the spectrum of 1,2-ethanediol 2-deoxy- α -D-*arabino*-hexopyranoside.¹³ The constants $^1J_{C(1'),H(1')} = 168.1$ and 167.9 Hz in the ^{13}C NMR spectra of compounds 10 and 11 also indicated the presence of an α -glycoside bond.¹²

Thus, glycosylation of primary and secondary hydroxyl groups in betulin monoacetates under conditions of acid catalysis stereoselectively gave 2-deoxy- α -hexopyranosides in high yields.

Experimental

TLC was carried out on Silufol plates (Chemapol, Czech Republic) using $CHCl_3$ -MeOH (20 : 1) as the eluent. Compounds were visualized using a 20% solution of phosphotung-

Table 1. ^{13}C NMR spectroscopis data for glycosides 7–16 (CDCl_3 , δ , J/Hz)*

Com- po- und	C(2)	C(3)	C(11)	C(20)	C(28)	C(29)	C(30)	C(1')	C(2')	C(3')	C(4')	C(5')	C(6')	OCOCH ₃	O ₂ COCH ₃
7	25.1	82.8	20.7	150.0	62.4	109.8	19.0	93.1	35.6	69.2	68.2	69.5	62.7	20.7, 20.8, 20.9, 20.9	169.9, 170.1, 170.6, 171.5
8	25.2	89.3	20.5	149.8	62.4	109.8	18.9	99.5	35.4	69.0	67.7	70.8	62.5	20.6, 20.7, 20.7, 20.9	169.7, 169.9, 170.1, 171.2
9	24.2	88.7	20.4	149.6	62.3	109.6	18.8	99.1	35.6	68.9	74.8	65.3	17.6	20.5, 20.6, 20.9, 21.5	169.8, 170.4, 171.0
10	23.6	80.7	20.6	150.1	66.4	109.7	19.0	97.7	35.0	69.1	67.9	69.3	62.1	20.7, 20.8, 21.1	169.7, 170.0, 170.6, 170.7
11	23.6	80.8	20.7	150.3	70.8	109.6	19.0	103.8	35.3	69.1	74.9	64.8	17.9	20.9, 21.2, 21.2	170.3, 170.8
12	24.7	81.7	20.3	149.9	60.5	108.8	19.1	92.7	36.5	71.4	68.5	72.1	62.5		
13	25.1	88.4	20.5	150.1	60.8	109.8	19.0	100.0	36.4	69.3	68.3	71.6	61.8		
14	25.1	88.7	20.8	150.4	60.4	109.6	19.0	100.0	39.0	69.2	78.1	67.5	17.5		
15	27.1	78.9	20.8	150.4	65.8	109.6	19.0	98.2	35.0	71.1	69.0	72.0	61.8		
16	27.1	79.0	20.8	150.5	65.6	109.7	19.1	97.8	35.0	69.3	78.0	67.6	17.8		

* $^1J_{\text{C}(1'),\text{H}(1')} = 169.5$ (7), 170.0 (8), 168.3 (9), 168.1 (10), 167.9 (11).

stic acid in EtOH followed by heating at 100–120 °C for 2–3 min. Column chromatography was performed on silica gel L (40/100 mm) (Chemapol, Czech Republic).

^{13}C and ^1H NMR spectra were recorded on a Bruker AM-300 spectrometer (75.5 and 300 MHz, respectively) in CDCl_3 using SiMe_4 as the internal standard. Melting points were determined on a Boetius hot micro stage.

MeCN and CH_2Cl_2 used for the syntheses were distilled two times over P_2O_5 . Molecular sieves (4 Å) were activated at 180–190 °C for 2 h (1–5 Torr). The KU-2-8 cationite (H^+ form) was dried by a reported procedure.⁴ 3-*O*-Acetylbetulin was synthesized by a procedure reported previously.¹⁴ Glycals 4–6 were synthesized using a known procedure.¹⁵

Extraction of betulin from the bark of the *Betula pendula* birch. A Pr^iOH –water mixture (7 : 3, 700 mL) containing dry crushed bark (70 g) was refluxed for 2 h. The bark was filtered off, and the hot solution was kept for 12 h at 0–5 °C. The precipitate that formed was filtered off, and the mother liquor was again used for extraction, which was repeated two times. The yield of dry extract was 24.7 g (35.3%). Betulin was obtained by chromatography of the extract using benzene as the eluent, yield ~75–80%, m.p. 251 °C.

28-Acetoxy-3 β -hydroxy-20(29)-lupene (2). A solution of betulin (4.42 g, 10 mmol) in 150 mL of glacial AcOH was refluxed for 1 h, concentrated to half the volume, and the residue was poured into 300 mL of cold water. The precipitate that formed was filtered off and washed to a neutral pH. The dry residue was chromatographed using benzene as the eluent. Yield 3.68 g (76%). M.p. 203–205 °C (cf. Ref. 16: m.p. 205 °C).

Synthesis of glycosides 7–11 (general procedure). Glycal 4 (0.27 g), 5 (0.27 g), or 6 (0.22 g), molecular sieves (0.3 g), anhydrous cationite (0.8 g), and LiBr (0.6 g) were added to a solution of betulin monoacetate 2 or 3 (1 mmol, 0.48 g) in 40 mL of a CH_2Cl_2 – CH_3CN mixture (1 : 1). The mixture was stirred for 4–5 h (TLC monitoring) and filtered. The filtrate was concentrated, and the residue was chromatographed using a heptane–AcOEt mixture, 7 : 1 \rightarrow 2 : 1, as the eluent.

The 5 : 1 \rightarrow 3 : 1 mixtures were used to obtain a TLC-homogeneous product.

28-Acetoxy-3 β -(3,4,6-tri-*O*-acetyl-2-deoxy- α -D-arabino-hexopyranosyloxy)-20(29)-lupene (7). Yield 0.59 g (78.3%). R_f 0.75, m.p. 171–173 °C. Found (%): C, 70.08; H, 9.31. $\text{C}_{44}\text{H}_{68}\text{O}_{10}$. Calculated (%): C, 69.81; H, 9.05. ^1H NMR, δ : 0.78, 0.80, 0.93, 0.95, 1.00 (all s, 15 H, 5 CH_3); 1.00–1.95 (m, 26 CH_2 , CH of aglycone, H(2')); 1.65 (s, 3 H, CH_3); 1.98, 2.01, 2.04, 2.06 (all s, 12 H, 4 Ac); 2.42 (ddd, 1 H, H(19), $J = 5.4, 11.7, 11.7$ Hz); 3.12 (dd, 1 H, H(3), $J = 5.8, 10.5$ Hz); 3.70–3.74 (m, 1 H, H(6')); 3.82 (d, 1 H, H(28), $J = 11.0$ Hz); 3.98–4.08 (m, 1 H, H(5')); 4.18–4.32 (m, 2 H, H(28), H(6')); 4.55 and 4.66 (both br. signals, 2 H, H(29)); 4.96 (t, 1 H, H(4')), $J = 9.9$ Hz); 5.11 (d, 1 H, H(1'), $J = 2.6$ Hz); 5.27 (ddd, 1 H, H(3'), $J = 5.2, 11.2, 9.9$ Hz).

28-Acetoxy-3 β -(3,4,6-tri-*O*-acetyl-2-deoxy- α -L-arabino-hexopyranosyloxy)-20(29)-lupene (8). Yield 0.61 g (80.1%). R_f 0.73, m.p. 105–107 °C. Found (%): C, 70.11; H, 8.95. $\text{C}_{44}\text{H}_{68}\text{O}_{10}$. Calculated (%): C, 69.81; H, 9.05. ^1H NMR, δ : 0.73, 0.77, 0.81, 0.90, 0.96 (all s, 15 H, 5 CH_3); 1.00–1.90 (m, 26 CH_2 , CH of aglycone, H(2')); 1.63 (s, 3 H, CH_3); 1.90, 1.96, 1.99 (all s, 12 H, 4 Ac); 2.38 (ddd, 1 H, H(19), $J = 5.5, 11.7, 11.7$ Hz); 2.98 (dd, 1 H, H(3), $J = 5.9, 10.6$ Hz); 3.78 (d, 1 H, H(28), $J = 11.0$ Hz); 3.97 (dd, 1 H, H(6'), $J = 2.0, 12.0$ Hz); 4.06 (ddd, 1 H, H(5'), $J = 9.8, 2.0, 6.2$ Hz); 4.19 (dd, 1 H, H(6'), $J = 6.2, 12.0$ Hz); 4.21 (d, 1 H, H(28), $J = 11.0$ Hz); 4.51 and 4.63 (both d, 2 H, H(29), $J = 2.0$ Hz); 4.88 (t, 1 H, H(4')), $J = 9.8$ Hz); 4.93 (d, 1 H, H(1'), $J = 2.8$ Hz); 5.25 (ddd, 1 H, H(3'), $J = 5.3, 11.8, 9.8$ Hz).

28-Acetoxy-3 β -(3,4-di-*O*-acetyl-2,6-dideoxy- α -L-arabino-hexopyranosyloxy)-20(29)-lupene (9). Yield 0.57 g (82%). R_f 0.77, m.p. 98–100 °C. Found (%): C, 72.45; H, 9.73. $\text{C}_{42}\text{H}_{66}\text{O}_8$. Calculated (%): C, 72.17; H, 9.51. ^1H NMR, δ : 0.75, 0.82, 0.88, 0.97, 1.05 (all s, 15 H, 5 CH_3); 1.00–2.15 (m, 26 CH_2 , CH of aglycone, H(2')); 1.29 (d, 3 H, H(6')), $J = 6.3$ Hz); 1.60 (s, 3 H, CH_3); 1.95, 1.96, 1.98 (all s, 9 H,

3 Ac); 2.32–2.40 (m, 1 H, H(19)); 2.93 (dd, 1 H, H(3), $J = 6.0, 10.1$ Hz); 3.77 (d, 1 H, H(28), $J = 11.0$ Hz); 3.96 (dq, 1 H, H(5'), $J = 9.7, 6.3$ Hz); 4.16 (d, 1 H, H(28), $J = 11.0$ Hz); 4.48–4.53 and 4.58–4.62 (both m, 2 H, H(29)); 4.64 (t, 1 H, H(4'), $J = 9.7$ Hz); 4.82 (d, 1 H, H(1'), $J = 2.1$ Hz); 5.20 (ddd, 1 H, H(3'), $J = 5.5, 11.7, 9.7$ Hz).

3 β -Acetoxy-28-(3,4,6-tri-*O*-acetyl-2-deoxy- α -D-arabino-hexopyranosyloxy)-20(29)-lupene (10). Yield 0.68 g (89.2%). R_f 0.72, m.p. 93–95 °C. Found (%): C, 70.13; H, 8.81. $C_{44}H_{68}O_{10}$. Calculated (%): C, 69.81; H, 9.05. 1H NMR, δ : 0.77, 0.78, 0.79, 0.92, 0.96 (all s, 15 H, 5 CH₃); 1.00–1.95 (m, 26 CH₂, CH of aglycone, H(2')); 1.62 (s, 3 H, CH₃); 1.95, 1.98, 1.99, 2.03 (all s, 12 H, 4 Ac); 2.35 (ddd, 1 H, H(19), $J = 5.5, 11.1, 11.1$ Hz); 2.96 and 3.80 (both d, 2 H, H(28), $J = 9.4$ Hz); 3.87 (ddd, 1 H, H(5'), $J = 9.7, 3.2, 2.1$ Hz); 3.96 (dd, 1 H, H(6'), $J = 2.1, 12.3$ Hz); 4.34 (dd, 1 H, H(6'), $J = 3.6, 12.3$ Hz); 4.41 (dd, 1 H, H(3), $J = 5.6, 10.1$ Hz); 4.52 and 4.62 (both d, 2 H, H(29), $J = 1.7$ Hz); 4.85 (d, 1 H, H(1'), $J = 2.8$ Hz); 4.96 (t, 1 H, H(4'), $J = 9.7$ Hz); 5.23 (ddd, 1 H, H(3'), $J = 5.2, 11.4, 9.7$ Hz).

3 β -Acetoxy-28-(3,4-di-*O*-acetyl-2,6-dideoxy- α -L-arabino-hexopyranosyloxy)-20(29)-lupene (11). Yield 0.63 g (90.1%). R_f 0.76, m.p. 111–113 °C. Found (%): C, 71.85; H, 9.24. $C_{42}H_{66}O_8$. Calculated (%): C, 72.17; H, 9.51. 1H NMR, δ : 0.76, 0.77, 0.90, 0.98 (all s, 15 H, 5 CH₃); 1.00–2.15 (m, 26 CH₂, CH of aglycone, H(2')); 1.15 (d, H(6'), $J = 6.3$ Hz); 1.61 (s, 3 H, CH₃); 1.96, 1.97, 2.01 (all s, 9 H, 3 Ac); 2.25–2.32 (m, 1 H, H(19)); 3.10 and 3.64 (both d, 2 H, H(28), $J = 9.5$ Hz); 3.73 (dq, 1 H, H(5'), $J = 9.2, 6.3$ Hz); 4.40 (dd, 1 H, H(3), $J = 5.3, 10.1$ Hz); 4.50 and 4.64 (both br. signals, 2 H, H(29)); 4.68 (t, 1 H, H(4'), $J = 9.2$ Hz); 4.85 (br. signal, 1 H, H(1')), 5.01 (ddd, 1 H, H(3'), $J = 5.5, 11.2, 9.2$ Hz).

Glycosides 7–11 were deacetylated with methanolic KOH by the reported procedure¹⁰ to give glycosides 12–16 in yields of 87–92%.

3 β -(2-Deoxy- α -D-arabino-hexopyranosyloxy)-28-hydroxy-20(29)-lupene (12). R_f 0.35, m.p. 170–172 °C. Found (%): C, 73.78; H, 9.95. $C_{36}H_{60}O_6$. Calculated (%): C, 73.42; H, 10.27.

3 β -(2-Deoxy- α -L-arabino-hexopyranosyloxy)-28-hydroxy-20(29)-lupene (13). R_f 0.33, m.p. 184–186 °C. Found (%): C, 70.21; H, 9.88. $C_{36}H_{60}O_6$. Calculated (%): C, 73.42; H, 10.27.

3 β -(2,6-Dideoxy- α -L-arabino-hexopyranosyloxy)-28-hydroxy-20(29)-lupene (14). R_f 0.36, m.p. 186–188 °C. Found (%): C, 75.85; H, 10.81. $C_{36}H_{60}O_5$. Calculated (%): C, 75.48; H, 10.56.

28-(2-Deoxy- α -D-arabino-hexopyranosyloxy)-3 β -hydroxy-20(29)-lupene (15). R_f 0.32, m.p. 145–147 °C. Found (%): C, 73.11; H, 10.61. $C_{36}H_{60}O_6$. Calculated (%): C, 73.42; H, 10.27.

28-(2,6-Dideoxy- α -L-arabino-hexopyranosyloxy)-3 β -hydroxy-20(29)-lupene (16). R_f 0.34, m.p. 168–170 °C. Found

(%): C, 75.75; H, 10.34. $C_{36}H_{60}O_5$. Calculated (%): C, 75.48; H, 10.56.

This study was financially supported by the Russian Foundation for Basic Research (Project No. 96-03-33240).

References

1. L. I. Filina and V. F. Semenchenko, *Mater. 49-i Region. konf. po farmatsii, farmakol. i podg. kadrov* [Materials of the 49th Regional conference on pharmacy, pharmacology, and training of personnel], Pyatigorsk, 1994, 78 (in Russian).
2. T. Fujioka, Y. Kashiwada, R. E. Kilkuskie, L. M. Cosentino, L. M. Ballas, J. B. Jiang, W. P. Janzen, I. S. Chen, and K. H. Lee, *J. Nat. Prod.*, 1994, 57, 243.
3. K. Yasukawa, Sy. Yu, S. Yamanouchi, M. Takido, T. Akihisa, and T. Tamura, *Phytomedicine*, 1995, 1, 309.
4. S. Sabesan and S. Neira, *J. Org. Chem.*, 1991, 56, 5468.
5. O. B. Flekhter, L. A. Baltina, E. V. Vasiljeva, and G. A. Tolstikov, *Mendeleev Commun.*, 1997, 3.
6. O. B. Flekhter, L. A. Baltina, and G. A. Tolstikov, *Izv. Akad. Nauk, Ser. Khim.*, 1997, 1390 [*Russ. Chem. Bull.*, 1997, 46, 1335 (Engl. Transl.)].
7. L. E. Odinkova, G. I. Oshitok V. A. Denisenko, V. F. Anufriev, A. M. Tolkach, and N. I. Uvarova, *Khim. Prirod. Soedin.*, 1984, 182 [*Chem. Nat. Compd.*, 1984 (Engl. Transl.)].
8. L. E. Odinkova, M. V. Denisenko, V. A. Denisenko, and N. I. Uvarova, *Khim. Prirod. Soedin.*, 1988, 212 [*Chem. Nat. Compd.*, 1988 (Engl. Transl.)].
9. L. F. Tietze, H. Heinzen, P. Moyna, M. Rischer, and H. Neunaber, *Liebigs Ann. Chem.*, 1991, 1245.
10. L. A. Baltina, O. B. Flekhter, E. V. Vasil'eva, and G. A. Tolstikov, *Izv. Akad. Nauk, Ser. Khim.*, 1996, 2340 [*Russ. Chem. Bull.*, 1996, 45, 2222 (Engl. Transl.)].
11. O. B. Flekhter, L. A. Baltina, E. V. Vasil'eva, and G. A. Tolstikov, *Izv. Akad. Nauk, Ser. Khim.*, 1996, 2993 [*Russ. Chem. Bull.*, 1996, 45, 2843 (Engl. Transl.)].
12. F. Barresi and O. Hindsgaul, *Synlett*, 1992, 759.
13. G. Descotes, H. Idmoumzet, and J.-P. Praly, *Carbohydr. Res.*, 1984, No. 128, 341.
14. K. Alb. Vesterberg and R. Vesterberg, *Arkiv. Kem., Mineral. Geol.*, 1926, 9, 17.
15. Yu. A. Zhdanov, G. N. Dorofeenko, G. A. Korol'chenko, and G. V. Bogdanova, *Praktikum po khimii uglevodov* [Practical course on carbohydrate chemistry], Vysshaya shkola, Moscow, 1973 (in Russian).
16. Yu. L. Yur'ev and V. I. Azarova, *Tez. dokl. III Vsesoyuz. nauchno-tekh. konf.* [Abstracts of the III All-Union Scientific-technical Conference], 15–18 May 1990, Gor'kii, 111 (in Russian).

Received July 30, 1997;
in revised form October 7, 1997